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Physico-chemical Analysis of Water of Railway Colony Area Pond, Chakradharpur, Near Waste Accumulated Area

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ABSTRACT

Waste is a refuse of the daily house hold activities beside other activities. The huge accumulation of waste in unscientific way causes a lot of problems. It gives bad smell, provide breeding place for mosquitoes and germs. The leachate from the waste dump deteriorates the underground water quality too. In the present communication physico-chemical analysis of water of railway colony area pond, chakradharpur, near waste accumulated area has been analysed. The same is the part of Ph.D. work to access the impact of waste dump to water of Chaibasa and chakradhrpur of West singhbhum, Jharkhand.

Key words: MSW, physico-chemical analysis, Chakradharpur.

INTRODUCTION

Waste is defined as discarded or unwanted materials without any value having no economic or aesthetic role for mankind. It may include such materials which have been treated as "disposed off" materials left over from industrial as well as other activities including household activities contaminating the surrounding atmosphere like air, water and soil. Its deposition is sometimes hazardous for mankind and other living systems. Therefore, waste is a word that refers to describe any material that is discarded or not in use any longer and needs to be disposed. Due to modernization and technological development, the solid waste generation has not only been increased but it has also changed the environmental make up causing a number of problems to human population as well as to aquatic lives. There has been arisen the situation of chronic poisoning in some of the localities causing adverse effect on terrestrial life. It also affect the under ground water quality.For analyzing the water quality the Chakradharpur, railway colony was selected as because of heavy population load and accumulation of MSW. The sampling was done in all the three seasons namely rainy, winter and summer.

Metallurgical Laboratory, Jamshedpur during summer month and in other seasons the analysis was done with the help of field water testing kit. The complete analytical work shows this sample to be far better with reference to concentration of pollutants and other harmful chemicals and salts. During all the seasons, water was more or less clear showing lesser concentration of dissolved solids and other organic materials. There was no sign of objectionable smell too.

PARAMETER	METHOD OF ANALYSIS				
рН	: pH Meter / pH paper				
Transparency	: Seechi Disc				
Turbidity	: Turbidometer				
Electrical Conductivity	: Conductivity meter /				
	Conductometer				
Dissolved Solids	: Conventional method				
Total Hardness	: Conventional method				
Sodium	: Ion Chromatograph /				
	Water Testing Kit				
Magnesium	: Ion Chromatograph /				
	Water Testing Kit				
Fluorine	: Ion Chromatograph /				
	Water Testing Kit				
Iron	: AAS / Water Testing Kit				
Nitrate, Phosphate, Sulphate : Water Testing Kit.					

MATERIALS AND METHOD

The analysis of water sample was done in National

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Result and Discussion :

The complete analytical work can be shown in the following table and graphical representation.

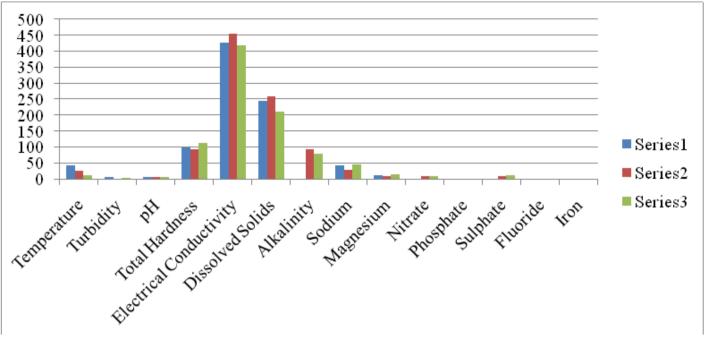
Sl. No.	PARAMETER	SUMMER	RAINY	WINTER
1.	Colour	Colourless	Slight Muddy	Colourless
2.	Odour	NIL	NIL	NIL
3.	Transparency	Clear	Slight clear	Clear
4.	Temperature	41.2	26.4	11.5
5.	Turbidity	5.2	1.2	3.4
6.	рН	6.5	6.1	6.4
7.	Total Hardness	100	94	112
8.	Electrical Conductivity	427	455	420
9.	Dissolved Solids	245	257	212
10.	Alkalinity	ND	94	80
11.	Sodium	43	28	44
12.	Magnesium	13	8	15
13.	Nitrate	ND	7.4	8.8
14.	Phosphate	ND	0.01	0
15.	Sulphate	ND	9.1	10.4
16.	Fluoride	0.4	0.36	0.49
17.	Iron	0.6	0.52	0.7

During analysis of turbidity, it was recorded 502 NTU during summer days, 1.2 NTU during rainy month and 3.4 NTU during winter. The pH value was totally up to mark showing no objectionable sign and it ranged from 6.1 to 6.5. Likewise, the estimation of total hardness showed it to be 100 mg/l during summer, 94 mg/l during rainy season and 112 mg/l in winter month.

The quantitative analysis of electrical conductivity shows that it was 427 µs/cm in summer; 455 µs/cm during rainy and 420 µs/cm during winter season. The estimated value of dissolved solids also showed more or less within range which was estimated 245 mg/l in summer, 258 mg/l in rainy and 212 mg/l in winter season. The alkalinity was not detected during analytical work at NML during summer season but during rainy season it was 94 mg/l and during winter, it was 80 mg/l.

The estimation of sodium salt showed its quantity to be 43 mg/l in summer, 28 mg/l in rainy and 44 mg/l in winter. Likewise, the concentration of magnesium salt was 13 mg/l during summer, 8 mg/l during rainy season and 15 mg/l during winter.

The analytical work of estimation of nitrate, phosphate and sulphate did not show their remarkable presence during summer months, but during rainy season the nitrate present was 7.4 mg/l; phosphate was 0.01 mg/l and sulphate was 9.1 mg/l. Likewise, during winter month, the concentration of nitrate was 8.8 mg/l; phosphate was nil and sulphate was 10.4 mg/l.





Finally, the analytical data of Fluorine shows its concentration in this water sample to be 0.4 mg/l in summer; 0.36 mg/l in rainy and 0.49 mg/l in winter season. The analysis of iron was also done as per above mentioned procedure and it was found to be 0.6 mg/l in summer month; 0.52 mg/l during rainy and 0.7 mg/l during winter season. The concentration of iron present in the sample was above the safe zone and it authenticates the higher concentration of iron concentration in West Singhbhum area along with the state of Jharkhand. References:

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